

① Phys 2426

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Lec 1

Why Study E & M?

- Our society is electric.
 - Energy
 - Info
 - Comm
- Light & Optics
- Chemistry
- Practice - Analysis & Abstraction

Electrostatics

All matter made of particles

Protons, Electrons, Neutrons,

Charge is the "value" of "how electric"

a particle or object is.

Charge (Q, q) is in coulombs (C).

Charges combine by addition.

Two types of charge, and they cancel.

Protons are \oplus

Electrons are \ominus

Charge is conserved.

The ~~Funda~~ SI Unit = 1.0 C is HUGE.

Note Uses of "e": e = "times ten to the power"

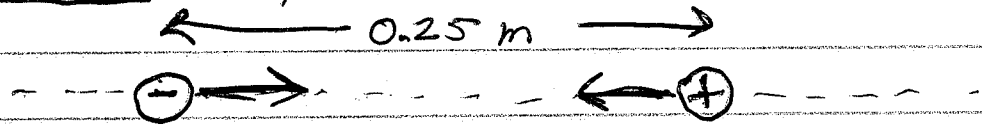
$$e = 1.6 \times 10^{-19} \text{ C}$$

$$e = 2.718 \dots$$

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Fundamental Interaction - Coulomb Force
Force is a vector: Magnitude & Direction

Direction Opposites Attract



$$q_1 = -5 \text{ nC}$$

$$q_2 = +8 \text{ nC}$$

Strength Coulomb's Law

$$k = 9 \times 10^9 \text{ Nm}^2/\text{C}^2$$

$$F_E = \frac{k q Q}{R^2}$$

q, Q = charge magnitudes

R = center-to-center dist

$$F_E = \frac{(9 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2})(5 \times 10^{-9} \text{ C})(8 \times 10^{-9} \text{ C})}{(0.25 \text{ m})^2}$$

$$= 5.76 \times 10^{-6} \text{ N} = 5.76 \mu\text{N}$$

What if one charge is zero? Zero force.

Can have "charge-induced dipole" attraction.

③

Accounting Methods

$$Q = q_1 + q_2 + \dots$$

↑
Total Charge

↑ individual charges

Ex: $q_p = (+1.6 \times 10^{-19} \text{ C}) = +e$
 $q_e = (-1.6 \times 10^{-19} \text{ C}) = -e$

$$Q = +e N_p - e N_e$$
$$= e (N_p - N_e)$$

Note $Q = 0 \rightarrow N_p = N_e$

Charge Density

$\rho =$ charge per volume $Q = \rho_{\text{avg}} V$

$$Q = \int \rho \, dV$$

$\lambda =$ charge per length $Q = \lambda_{\text{avg}} l$

$$Q = \int \lambda \, dx$$

$\sigma =$ charge per area $Q = \sigma_{\text{avg}} A$

$$Q = \int \sigma \, dA$$

$$Q = \int \sigma \, dA$$